



IJSRM

INTERNATIONAL JOURNAL OF SCIENCE AND RESEARCH METHODOLOGY

An Official Publication of Human Journals



Human Journals

Review Article

October 2016 Vol.:4, Issue:4

© All rights are reserved by Dr. D. Sarvamangala et al.

Production of Gallic acid-A Short Review



IJSRM

INTERNATIONAL JOURNAL OF SCIENCE AND RESEARCH METHODOLOGY

An Official Publication of Human Journals



Ch. Raghu Babu¹, N. Sowjanya², Dr. D. Sarvamangala*

*1&2 2/4 B.Tech Biotechnology, *Associate professor, GIT,
GITAM University, India*

Submission: 3 October 2016

Accepted: 10 October 2016

Published: 25 October 2016

Keywords: Gallic acid, trihydroxybenzoic acid, phenolic acid, 3,4,5-trihydroxybenzoic acid, gallate

ABSTRACT

Gallic acid is a trihydroxybenzoic acid, a type of phenolic acid, known as 3,4,5-trihydroxybenzoic acid also known as gallate. Literature survey reveals that fungal species of *A. niger* were extensively used by many workers for the production of Gallic acid.



HUMAN JOURNALS

www.ijsrm.humanjournals.com

HUMAN

INTRODUCTION

Gallic acid is an organic acid found in a variety of foods and herbs that are well known as a powerful antioxidant. The chemical formula is $C_6H_2(OH)_3COOH$.

Gallic acid is found both free and as part of tannins. Salts and esters of gallic acid are termed gallates¹. It does not contain gallium though its name gallic acid. It is an endogenous product found in plants. Gallic acid is one of the active compounds that have potent anti-angiogenic and alpha-glucosidase inhibitory activities.

Gallic acid is commonly used in the pharmaceutical industry because many *in vivo* and *in vitro* studies in humans, animals, and cell culture have provided evidence for the following actions of gallic acid. It shows cytotoxicity against cancer cells. It can be used to treat albuminuria and diabetes. It seems to have antifungal and antiviral properties, used as an antioxidant and helps to protect human cells against oxidative damages. It can be used as a remote astringent in cases of internal hemorrhage used to treat psoriasis and external hemorrhoids containing gallic acid^{3,4}. It is used in brewing industry and wine industry as clarifying agent and as flavoring agent in foods.

Pharmacological investigations have shown that this gallic acid has several biological activities such as antimicrobial, antifungal and antiviral, anti-inflammatory, antioxidant, anticarcinogenic and antimutagenic activities. Gallic acid is toxic to animals because of its reaction with DNA or proteoglycans.

Gallic is used in printing inks. Gallic acid is compound of interest to both pharmaceutical and chemical industries because of its several interesting properties and commercial applications⁵. Gallic acid is also found in gallnuts, sumav, witch hazel, watercress, oak bark, tea leaves, areca nut, bearberry (*Arctostaphylos*), blackberry, *Caesalpinia mimosoides*.

Properties

Pure gallic acid is a colorless crystalline organic powder, while salts and esters of gallic acid are termed gallate.

Physical and Chemical Properties²

Appearance: Fine crystals, white yellowish-white or pale, fawn-color.

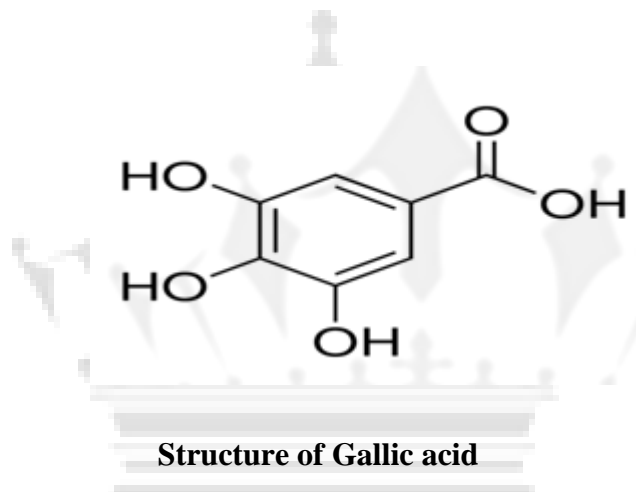
Odor: Odorless.

Solubility: 1.1g/100ml water @ 20⁰C (68F) (anhydrous).

Density: 1.7 (anhydrous).

% Volatiles by volume @ 21C (70F): 1

Melting Point: 250⁰C (482F).



Uses of Gallic Acid

Ancient applications of Gallic acid:

Ancient cultures often believed in remedies and medicines that have later been moved to have ineffective. Foods and herbs rich in gallic acid usage in ancient cultures were not entirely inaccurate^{5,6}. In actuality, these medicines and treatments often did work.

Past uses of Gallic acid:

Many of the foods containing Gallic acid have been used for years as natural remedies and were relied upon by various cultures for their medicinal properties. For example: Native Americans and early American settlers used the Blueberries to make an aromatic tea that was used as a relaxant during childbirth.

The list could continue for each of the foods where Gallic acid is present, for ancient cultures quickly learned of the beneficial properties of the plants and herbs that surrounded them, and many herbs and healthy foods contain Gallic acid^{7,8}. The ancient cultures may not have understood the role of gallic acid even that gallic acid existed, in the foods and herbs they used for medicinal purposes. They did, however, believe that their medicines worked.

Modern uses of Gallic acid:

Gallic acid is believed to have health benefits to those at risk for certain forms of cancer or neural disorders. Studies have shown that neuronal death can be inhibited by gallic acid and it has anti-cancer properties against leukemia, certain prostate, colon and lung cancer cells^{9,10}. Gallic acid has been shown to prevent cellular mutations and to be toxic to cancer cells while having no negative effect on healthy cells.

Gallic acid has antiviral and antifungal properties. It is a powerful antioxidant that helps to prevent oxidative damage^{11,12,13}. Gallic acid also has therapeutic applications for inflammatory allergic diseases, such as asthma, allergic rhinitis, sinusitis due to its ability to inhibit histamine release and the expression of pro-inflammatory cytokine.

Clinical Pharmacology:

Gallic acid is a member of the family of chemicals called tannins. In much of literature, the tannins are taught to be responsible for the survival of higher plants over earlier less developed plant species. Gallic acid's role in those plants is significant as a pesticide. Digestibility of insects decreases which eat leaves containing tannins. Gallic acid has been beneficial in muscular skeleton problems, particularly low back pain. As with many phenolics, it benefits chronic fatigue.

A study was conducted on acidum gallicum tablets that contained 10% GA and 90% glucose and a black tea brew that contained 93% of its GA in free form to determine the pharmacokinetics and relative bioavailability of GA in healthy humans^{14,15}. Concentrations of GA and its metabolite, 4-O-methylgallic acid (4OMGA), were determined. GA from both the tablets and tea was rapidly absorbed and eliminated GA concentrations in the stomach could achieve a maximum of 1.5 mmol/L (there was 0.3 mol GA in 200ml tea brew) highest concentration observed in plasma^{16,17}.

GA has a dual role as antimelanogenic and antioxidative agents and effective compound for skin health. The putative effect of GA on the inhibition of skin tyrosinase and melanogenesis under *in vivo* conditions based our current *in vitro* data^{18,19}. Gallic acid derived molecules can also act as plant growth promoters^{20,21}.

Forms of Gallic acid:

There are not supplements specifically of Gallic acid. However, there are plenty of readily available vitamins and nutritional supplements that contain Gallic acid and will help to give the benefits associated with Gallic acid. Most of these supplements are specifically designed to be rich antioxidants²².

Some examples of pill from supplements that include Gallic acid are grape seed extract, rooibos tea extract, among others^{23,24}.

From natural juices, we can increase the intake of Gallic acid such as pomegranate, blueberry and grape juice, or through foods rich in Gallic acid²⁵.

Vitamins supplements containing gallic acid in pill form are available at local health stores. Prices vary depending on the brand purchased.

Side effects of Gallic acid:

Gallic acid has triggered contractile responses and inhibited the decrease of vascular pressure in the thoracic aorta. Gallic acid also interferes with the medications taken for high blood pressure or for heart problems²⁶. Its use as an antioxidant helps to ensure overall and continued health and can be taken as a supplement to increase the levels of this beneficial antioxidant²⁷.

Table 1. Microorganisms and Raw materials used for the production of gallic acid

Sr. no	Name of the organism	Raw materials	Duration	Yield	Reference	Year
1	<i>Aspergillus niger</i>	Tara fruit pods	Time/h :- 45.0	30%	33	1985
2	<i>Aspergillus</i> sps	Walnut	—	83.3%	44	1987
3	<i>Aspergillus niger</i>	Sumac leaves	Time/h:- 40.0	9.75%	34	1987
4	<i>Aspergillus niger</i>	Gall nut	Time/h:- 24.0	40.5%	35	1989
5	<i>Aspergillus</i> and <i>P. zancintha</i>	Tara fruit pods	Time/h:- 30.0	30%	41	1990
6	<i>Rhizopus Oryzae</i> (free cells)	2% tannic acid in media	Time/h:- 96.0	83.5%	36	1997
7	<i>Rhizopus oryzae</i> (immobilized cells)	2% tannic acid in media	Time/h:- 96.0	78.5%	37	1997
8	<i>Aspergillus niger</i>	<i>Larrea tridentata</i>	—	39.4%	46	1997
9	<i>Rhizopus oryzae</i>	Teri pod cover	Time/h:- 72.0	90.90 %	30	1999
10	<i>Aspergillus fischer</i> , <i>Aspergillus niger</i> and <i>P. spinulosum</i>	<i>Quercus infectoria</i> gall nuts	—	91.3% 93.2%	43	2003
11	<i>Rhizopus oryzae</i>	Myrobalan and teri pod cover (mixed substrate)	Time/h:- 60.0	85.67 %	28	2004
12	<i>Aspergillus oryzae</i>	solid cashew husk	Time/h:- 48.0	0.14. %	29	2004
13	<i>Aspergillus foetides</i>	Myrobalan and teri pod cover (mixed substrate)	Time/h:- 72.0	90.48 %	38	2004
14	<i>Rhizopus oryzae</i> and <i>Aspergillus foetidus</i> (co-culture)	Myrobalan and teri pod cover (mixed substrate)	Time/h:- 48.0	94.8%	39	2005
15	<i>Aspergillus</i> and <i>P. zancintha</i>	Tara fruit pods	—	46.1%	42	2005
16	<i>Aspergillus</i> sps	Walnut	—	96%	45	2005
17	<i>Aspergillus niger</i>	Solid	Time/h:- 43.0	15%	31	2007
18	<i>Aspergillus niger</i>	Terminalia	—	72%	32	2007

		chebula				
19	<i>Aspergillus awamori</i>	Tannase	—	25.28 1 ug/ml	40	2012
20	<i>Aspergillus niger</i>	<i>Mimusops elengi</i>	Time(days):6-12 Temp: 25-45 ⁰ c pH: 4-9	46%	47	2015

RESULTS AND DISCUSSION

The data reveals that *A. niger* species is widely used for the production of Gallic acid and Walnuts were found to be the best source of production. Since Gallic acid is having several medicinal properties, there is lot of scope for tapping other cheap sources containing tannins.

REFERENCES

- 1) Heggerman A E and Butler L G, J. Agric. Food. Chem, 1978, 26, 809-812.
- 2) H. Chiu-Lan, L. Yuh-Charn, Y. Gow-Chin, C. Yui-Yin, Food Chem, 103 (2007) 528.
- 3) Singleton, V. L. (1981) Naturally occurring food toxicants: phenolic substances of plant origin common in foods. Adv. Food Res. 27: 149-242.
- 4) Bajpai, B.; Patil, S. (1996). Tannin acyl hydrolase (EC 3.1.1.20) activity of *Aspergillus*, *Penicillium*, *Fusarium* and *Trichoderma*. W.
- 5) Hearing V. J., J. Dermatol. Sci., 37, 13—14 (2005).
- 6) Briganti S., Camera E., Picardo M., Pigment Cell Res., 16, 101__110 (2003).
- 7) Phytochemicals-[http:// www.phytochemicals.info/phytochemicals/gallicacid.php](http://www.phytochemicals.info/phytochemicals/gallicacid.php).
- 8) Felter, H. W., Lloyd, J. U. (1898). *Acidium gallicum* (U. S. P)-gallic acid. [www.henriettesherbal.com/ eclectic/kings/acidum-gall.html](http://www.henriettesherbal.com/electic/kings/acidum-gall.html).
- 9) Mirvish, S. S, CRDESA, A., WALLCAVE, L., & SHUBIK, P. (1975). Induction of mouse lung adenomas by amines or ureas plus nitrite and by nitroso compounds: effect of ascorbate, gallic acid, thiocynate and caffeine. Journal of the National Cancer Institute, 55, 633-636.
- 10) Bradfield, A. E. & Bate-Smith, E. C. (1950). Biochim. Biophys. Acta, 4, 441.
- 11) Tack, B. F., P. J. Chapman, and S. Dagley. 1972.
- 12) Metabolism of gallic acid syringic acids by *Pseudomonas putida*. J. Biol. chem. 247:6438-6443.
- 13) Sparings, V. L., P. J. Chapman, and S. Dagley. 1974. Bacterial degradation of 4-hydroxyphenylacetic acid and homoprotocatechuic acid. J. Bacteriol. 120:159-167.
- 14) Lu Z., Nie G., Belton P. S., Tang H., Zhao B., Neurochem. Int., 48, 263—274(2006).
- 15) Aissam, H.; Errachidi, F.; Penninckx, M.; Merzouki, M.; Benlemilh, M. (2005). Production of tannase by *Aspergillus niger* HA37 growing.
- 16) H. Aissam, F. Errachidi, M. J. Penninckx, Email author, M. Merzouki, M. Benlemilh. Production of tannase by *Aspergillus niger* HA37 growing on tannic acid and Olive Mill Waste Waters. W. J. Microbial, Biotechnol., 21 (4), 609-614. (2005).
- 17) Batra, A.; Saxena, R.K. (2005). Potential tannase producers from the genera *Aspergillus* and *Penicillium*. Process Biochem., 40 (5), 1553-1557 32,24-26.
- 18) Solar, A., Colaric, M., usenik, V. and Stampar, F. 2006. Seasonal variations of selected flavonoids, phenolic acids and quinines in annual shoots of common walnut (*Junglans regia L.*). Plant Science 170:453-461.
- 19) Watanabe, A. Studies on the metabolism of gallic acid by microorganisms. Part 3. On the intermediary metabolism of gallic acid by *Aspergillus niger*. Agr. Biol. Chem., 29, 20-26. (1965).

- 20) Schindhelm RK, Van Der Zwan LP, Teerlink T, Scheffer PG (2009).
- 21) FENG Ding-kun et al, Journal of Anhui Agricultural Sciences, 2010-20.
- 22) Booth, A. N., Masri, M.S., Robbins, D. J., Emerson, O.H., Jones, F. T. & Deeds, F. (1959) The metabolic fate of gallic acid and related compounds. J. Biol. Chem. 234: 3014-3016. Watanabe, A. & Oshima, Y. (1965) Metabolism of gallic acid and tea catechin by rabbit. Agric. Biol. Chem. 29: 90-93.
- 23) Keller, K., Greimer, S. & Stockbrand, P. (1995) Homoeopathische Arzneimittel. Eschborn: Govi, Germant (monograph).
- 24) Wang, Z. Y., Huang, M. T., Lou, Y. R., Xie, J. G., Reuhl, K. R., newmark, H. L., Ho, C. T., Yang, C. S. & Conney, A. H. (1994) Inhibitory effects of black tea, green tea, decaffeinated black tea, and decaffeinated green tea on ultraviolet B light-initiated SKH-1 mice. Cancer Res. 54: 3428-3435.
- 25) Wang, Z. Y., Hong, J. Y., Huang, M. T., Reuhal, K. R., Conney, A. Hyang, C. s. (1992) Inhibition of N-nitrosodiethylamine and 4-(methylnitrosamin0)-1-(3-pyridyl)-1-butanone-induced tumorigenesis in A/J mice by green tea and black tea. Cancer Res. 52: 1943-1947.
- 26) Kim Y. J., No J. K., Lee J. H., Chung H. Y., Biol. Phram. Bull., 28, 323—327(2004).
- 27) Nerya O., Vaya J., Musa R., Izrael S., Ben-Arie R., Tamir S., Agric. Food chem., 51,1201—1207(2003).
- 28) Mukherjee G, Banerjee R. J. Basic. Microbiol.2004, 44:42-48.
- 29) Belmares-Cerda RE. Composition and fungal biodegradation of some polyphenolic compounds present in the medical semi arid region, Msc thesis, Universidad Autonomy de Coahulia, Mexico, 2004.7-30.
- 30) Kar B,Banerjee R, Bhattacharya BC. Ind. J. Microbiol. Biotechnol, 1999, 23:173-177.
- 31) Trevino-Cueto B, Luis M, Contreras-Esquivel JC,et al. Bioresour. Technol.2007, 98:721-724.
- 32) N. Lokeswari and K. Jayaraju. ISSN:0973-4945; CODEN ECJHAO E-Journal of chemistry vol.4, No 2,pp 287-293,2007.
- 33) Pourrat. H, Regerat F, Pourrat A, Daniel J (1985). Production of gallic acid from tara by a strain of *Aspergillus niger*. J. Ferment. Technol.63: 401-403. Sariozlu and Kivanc 1115.
- 34) Pourrat H. Regerat F, Pourrat A, Morvan P.(1987). Microbiological production of gallic acid from *Rhus Coriaria* L. Biotechnol. LeH. 9:731-734.
- 35) Regerat, F.; Pourrat H.; Pourrat, A. Journal of the American Leather Chemists Association, Volume Number: 84 , Page Numbers: 323-328, 1989.
- 36) Misro, S., Kumar, M., Banerjee, R. et al. Bioprocess Engineering, 16: 257. (1997).
- 37) Misro, S., Kumar, M., Banerjee, R. et al. Bioprocess Engineering, 16: 257. (1997).
- 38) Gargi Mukherjee, Rintu Banerjee, Journal of Basic Microbiology, Volume 44, Issue 1 Pages 42–48, 2004.
- 39) Rintu Banerjee, Gargi Mukherjee, Krushna Chandra Patra, Bioresource Technology Volume 96, Issue 8, Pages 949–953. 2005.
- 40) Arafat B.EL-TANASH, Abdel-Dayem A. SHERIEF and Alshaymaa NOUR (Vol.10, Issue of march, 2012, received october2,2012/ Accepted January 10,2012.) Botany Department, Faculty of science, Mansoura University, EGYPT.
- 41) Vermeire, A.; Vandamme, E.(1990).Fungal conversion of gallotannins into gallic acid. Ferment. Technol. Ind. Appl., 198-203.
- 42) A., Murray, T., Ward, E., Samuel, A., Tiwari, R. C., Glamor, A., Feuer, E. J. and tun, M. J. (2005) cancer statistics CA. Cancer J. clin., 55,10-30.
- 43) Gomes, C. A., de Cruz, T.G., Andrade, J.L., Milhazes, N., Borges, F. and marques, M.P. (2003). Anticancer activity of phenolic acids of natural (or) synthesis origin: a structure-activity study. J. med. chem., 46, 5395-5401.
- 44) Pourrat H. Regerat F, Pourrat A, Morvan P. (1987). Microbiological production of gallic acid from *Rhus Coriaria* L. Biotechnol. LeH. 9:731-734.
- 45) Mahapatra K, Nanda RK, Bag SS, Banerjee R, Pandey A, Szakacs G(2005). Purification, Characterization and some stdies on secondary structure of tannase from *Aspergillus Awamori Nakazawa*. Process biochem.40: 3251-3254.
- 46) Misro SK, Kumar MR, Banerjee R, Bhattacharyya BC (1997). Production of gallic acid by immobilization of *Rhizopus oryzae* Bioprocess Eng. 16: 257-260.
- 47) Proceedings of the 5th annual International Conference on Advances in Biotechnology (Biotech 2015), ISSN- 2251 2489, Published by Global Science and Technology Fourm (GSTF) 2015.